THE CHEMISTRY OF VICINAL TRICARBONYLS SYNTHESIS OF 3,4,7-TRISUBSTITUTED CARBACEPHAMS

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Abstract: Carbapenams substituted at the 3,4, and 7 positions have been prepared, starting with an intramolecular addition of a β -lactam to the central carbonyl of a tricarbonyl residue. The stereochemistry of the fused ring system was elucidated by X-ray analysis.

In earlier work,¹ we have explored the use of the highly electrophilic central carbonyl group of a tricarbonyl system to form carbapenam and carbacepham systems, as illustrated in the conversion of 1 to cepham 2 of undetermined stereochemistry.^{1a} We now report further studies on the use of this methodology for the formation of cephams, in which substituents are incorporated at the 3,4 and 7 positions, and all stereochemical features are elucidated.

Our synthesis began with the dioxinone 4 previously prepared by Bodurow² from the β -lactam 3, with clearly established *cis*-stereochemistry. Treatment of 4 with LDA and TBDMSCl in THF at -78°C afforded the expected N-silylated product 5 (83%) which, on reaction with benzyl alcohol, provided the N-protected β -keto ester 6 (94%).

Ester 6 was then converted to enamine 7 by reaction with dimethylformamide dimethyl acetal (DMF-DMA), followed by oxidative cleavage with ozone¹ yielding the vicinal tricarbonyl derivative 8 (79%). Initial attempts to effect desilylation with TBAF resulted in substantial decomposition along with a small amount of the fused ring product (16%). However, desilylation with HF-pyridine complex³ followed by treatment with silica gel afforded the cyclized product 9 in 54% yield. The structure of 9 was confirmed by X-ray crystallographic analysis (Figure 1),⁴ showing the α -orientation of the hydroxyl group at C4.

Figure 1. ORTEP drawing of cepham 9.

Reduction of 9 with sodium triacetoxyborohydride in glacial acetic acid gave, exclusively, diol 10 (92%), which underwent hydrogenolysis over palladium on activated carbon to provide the free acid 11 (90%).5 Acetylation of diol 10 yielded diacetate 12 (91%) convertible to carboxylic acid 13 (93%) on hydrogenolysis.

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- (1) (a) Wasserman, H. H.; Han, W. T. Tetrahedron Lett. 1984, 25, 3743 and 3747. (b) Wasserman, H.H.; Han, W. T. J. Am. Chem. Soc. 1985, 107, 1444. (c) Wasserman, H. H.; Henke, S. L.; Luce, P.; Nakanishi, E. J. Org. Chem. 1990, 55, 5821.
- (2) (a) Bodurow, C.; Carr, M. A. Tetrahedron Lett. 1989, 30, 4081. (b) Bodurow, C.; Carr, M. A.; Moore, L. L. Org. Prep. Proced. Int. 1990, 22, 109.
 (3) Nicolaou, K. C.; Seitz, S. P.; Pavia, M. R. J. Am. Chem. Soc. 1981, 103, 1222.
- (4) For compound 9: $C_{23}H_{18}O_7N_2$, MW 434, monoclinic, colorless crystals, space group with a = 10.092 (1) Å, b = 17.013 (2) Å, c = 12.3911 (9) Å, b = 104.529 (7)°, V = 2059.5 (3)Å³, z = 4, p = 1.401 g/cm³. Data were collected at 23°C and the diffraction measurements were made on a four-circle Rigaku AFC5S fully automated diffractometer using graphite monochromated $CuKa(\lambda = 1.54178 \text{ Å})$.
- (5) Assignment of the diaxial trans-relationship of the alcohol residues at the 3- and 4-positions is in accord with the expected mode of hydride delivery to the C-3 ketone from the reducing agent complexed with the C-4 hydroxyl.6
- (6) (a) Saksena, A. K.; Mangiaracina, P. Tetrahedron Lett. 1983, 24, 273. (b) Turnbull, M. D.; Hatter, G.; Ledgerwood, D. E. Tetrahedron Lett. 1984, 25, 5449.